



Studying invitro antioxidant effect of different extract of *Premna tomentosa* (Linn) aerial parts

Kottai Muthu A^{*1}, Dinesh Babu J²

¹Department of Pharmacy, Annamalai University, Annamalai Nagar, Chidambaram-608 002, Tamilnadu, India

²Scient Institute of Pharmacy, Khanapur Village, Ibrahimpatnam, Rangareddy - 501506, Telangana, India

Article History:

Received on: 08 Dec 2019

Revised on: 20 Jan 2020

Accepted on: 22 Jan 2020

Keywords:

Methanolic concentrate, *Premna tomentosa*, Superoxide radical, hydroxyl radical

ABSTRACT

The genus *Premna* (Verbenaceae) comprises a group of more than 200 different trees, distributed in tropical and subtropical areas of the world. *P. tomentosa* (Verbenaceae) is a well-known medicinal plant used extensively for the treatment of various ailments. In the present study, the entire plant of different concentrates of aerial parts of *Premna tomentosa* was evaluated for its *in-vitro* anti-oxidant potential by superoxide scavenging activity, hydroxyl radical scavenging taking quercetin and ascorbate as the standard correspondingly. The results of antioxidant activity superoxide radical activity and hydroxyl potential were expressed in terms of % inhibition of generated free radicals respectively with respect to various concentrations. An IC₅₀ value was recorded; methanolic concentrate of *Premna tomentosa* is more effective in superoxide radical activity, hydroxyl radical than that of EA and petroleum ether concentrate. The methanolic concentrate of *Premna tomentosa* and standard exhibited radical scavenging activity possessing IC₅₀ values 225 μg/mL and 60 μg/mL (Superoxide Scavenging Activity), 288 μg/mL and 65 μg/mL (hydroxyl radical activity) correspondingly. All the above invitro studies clearly indicate that the methanolic concentrate of *Premna tomentosa* has better free radical scavenging activity. These *invitro* estimations point out that this methanolic concentrate of *Premna tomentosa* is a paramount source of expected antioxidant, which may be supportive in preventing the improvement of a variety of free radical induced diseases.



*Corresponding Author

Name: Kottai Muthu A

Phone: 9443171712

Email: akottaimuthu@gmail.com

ISSN: 0975-7538

DOI: <https://doi.org/10.26452/ijrps.v11i2.2093>

Production and Hosted by

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INTRODUCTION

Free radicals induce lipid peroxidation and provokes damage in cell membrane. Incorporation of antioxidants in fats and oils or foods can prevent the deterioration of lipids in foods. Some of these compounds which retard lipid peroxidation are synthetic antioxidants whereas others occur as natural dietary constituents (Duh, 1999). The human body is in constant battle to keep from free radicals. The first line of defence is the preventive antioxidants which quench the free radicals generated in the body. An early stage Augmentation of antioxidant status should either prevent or greatly curtail tissue injury (Lobo *et al.*, 2010). Due to the adverse effects

of synthetic antioxidants on human health, restrictions have been imposed on their usage. Natural antioxidants have been greatly intensified as synthetic antioxidants exert carcinogenic effect (Bajpai *et al.*, 2014). Plants are enriched with bioactive ingredients which have most of the free radical scavenging and therapeutic function.

Premna tomentosa (Verbanaceae) is a glowing recognized herb used broadly for the cure of different disorders. *Premna tomentosa* was used for the cure the different disorders in India (Kubitzki and Kadereit, 2005). *Premna tomentosa* bark concentrate was used for treatment of hepatic disorders. *Premna tomentosa* leaves concentrates were used as diuretic (Krishnamurthi, 1969), *Premna tomentosa* was used for the treatment of lipid lowering (Devi *et al.*, 2004), antioxidant (Devi *et al.*, 1998), immune modulatory activities (Devi *et al.*, 2003) and mitochondrial dysfunction properties against acetaminophen induced rats (Devi *et al.*, 2005). Flavonoids, triterpenoids and steroids were isolated from *Premna tomentosa* (Chin *et al.*, 2006; Alam *et al.*, 1993). Still, no literature are available on the antioxidant activity of *Premna tomentosa* aerial parts. Thus, the present study to assess antioxidant activities of *Premna tomentosa* aerial parts.

MATERIALS AND METHODS

Gathering & Identification of Plant

The aerial parts of *Premna tomentosa* (family Verbanaceae) were gathered from Chenkottai, Tirunelveli District of Tamilnadu India. Plant recognition was made from Botanical investigation of India, Palayamkottai The *Premna tomentosa* were desiccated under shadowy, segregate, crushed through grinder (Sivakrishnan and KottaiMuthu, 2014).

Preparation of Concentrates

The pulverized materials were packed in muslin cloth and concentrated with pet.ether, ethyl acetate and methanol as solvents respectively according to the increasing order of polarity (Shajiselvin *et al.*, 2010) through hot constant percolation method in Soxhlet equipment (Harborne, 1984) for twenty-four hours. The concentrates were concentrated through rotational evaporator and subjected to solidify drying till dry powder was acquired by using lyophilizer. (SatheeshKumar *et al.*, 2010; Alagumanivasagam *et al.*, 2012).

Assessment of Antioxidant potential through invitro methods

The variety of concentrate of *Premna tomentosa* were used as assessment of antioxidant activity by Winterbourn *et al.* (1975) method described for

Superoxide radical (O_2^-) assay and Kunchandy and Rao (1990) method was adopted to determine the OH radical assay.

RESULTS AND DISCUSSION

Superoxide anion activity

Super-oxides might be produced in huge amounts by a variety of biological processes. It is recognized to be more harmful to cellular components as creator of the generally ROS, contributing to tissue injure and various disorders (Halliwell and Gutteridge, 1999). Superoxide radical potential of PE concentrates of *Premna tomentosa* appeared in Table 1. The more activity of PE concentrates of *Premna tomentosa* and quercetin at 1000 μ g/mL were recorded to be 57.68% & 99.12% correspondingly. The IC₅₀ of PE concentrates of *Premna tomentosa* & quercetin were recorded as 762 μ g/mL and 60 μ g/mL correspondingly.

Superoxide radical potential of EA concentrates of *Premna tomentosa* appeared in Table 2. The more activity of EA concentrates of *Premna tomentosa* and standard at 1000 μ g/mL was found to be 75.66% and 99.12% correspondingly. IC₅₀ EA concentrates of *Premna tomentosa* and standard was recorded as 375 μ g/mL and 60 μ g/mL correspondingly.

Superoxide radical potential of Methanol concentrates of *Premna tomentosa* appeared in Table 3. The more activity of Methanol concentrates of *Premna tomentosa* and standard at 1000 μ g/mL was found to be 78.68% & 99.12% correspondingly. The IC₅₀ EA concentrates of *Premna tomentosa* and standard was recorded as 225 μ g/mL and 60 μ g/mL correspondingly.

IC₅₀ values and percentage scavenging capacity, it was recorded that methanolic concentrates of *Premna tomentosa* had better activity in superoxide radical when compared EA and PE concentrates.

Hydroxyl radical scavenging activity

Hydroxyl radical activity was determined by generating the OH radicals using ascorbic acid. OH radicals were produced by the oxidation reaction with the Dimethylsulfoxide to give in HCHO, which provides an appropriate method to identify OH radicals by added with Nash reagent (Pavithra and Vadivukkarasi, 2015).

Hydroxyl radical activity of PE concentrate of *Premna tomentosa* appeared in Table 4. The hydroxyl radical activity of PE concentrate of *Premna tomentosa* and ascorbic acid at 1000 μ g/mL was found to be 49.55% & 96.50 % correspondingly. The IC₅₀ of PE concentrate of *Premna*

Table 1: Activity of PE concentrate of *Premna tomentosa* on Superoxide radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		PE concentrate	Quercetin
1	125	29.56 \pm 0.024	72.76 \pm 0.012
2	250	42.62 \pm 0.038	90.29 \pm 0.014
3	500	48.46 \pm 0.052	96.89 \pm 0.010
4	1000	57.68 \pm 0.069	99.12 \pm 0.018
		IC ₅₀ = 762 $\mu\text{g}/\text{mL}$	IC ₅₀ = 60 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

Table 2: Activity of EA concentrate of *Premna tomentosa* on superoxide radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		EA concentrate	Quercetin
1	125	23.54 \pm 0.024	72.76 \pm 0.012
2	250	37.25 \pm 0.045	90.29 \pm 0.014
3	500	56.80 \pm 0.052	96.89 \pm 0.010
4	1000	75.66 \pm 0.034	99.12 \pm 0.018
		IC ₅₀ = 375 $\mu\text{g}/\text{mL}$	IC ₅₀ = 60 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

Table 3: Activity of methanol concentrate of *Premna tomentosa* on Superoxide radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		Methanolic concentrate	Quercetin
1	125	40.34 \pm 0.045	72.76 \pm 0.012
2	250	54.42 \pm 0.056	90.29 \pm 0.014
3	500	69.44 \pm 0.031	96.89 \pm 0.010
4	1000	78.68 \pm 0.022	99.12 \pm 0.018
		IC ₅₀ = 225 $\mu\text{g}/\text{mL}$	IC ₅₀ = 60 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

Table 4: Activity of PE concentrate of *Premna tomentosa* hydroxyl radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		PE concentrate	Ascorbate
1	125	20.22 \pm 0.064	70.34 \pm 0.38
2	250	29.22 \pm 0.034	85.54 \pm 0.15
3	500	38.68 \pm 0.022	90.23 \pm 0.24
4	1000	49.55 \pm 0.064	96.50 \pm 0.16
		IC ₅₀ = 1038 $\mu\text{g}/\text{mL}$	IC ₅₀ = 65 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

Table 5: Activity of EA concentrate of *Premna tomentosa* hydroxyl radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		EA concentrate	(Ascorbate)
1	125	20.32 \pm 0.045	70.34 \pm 0.38
2	250	44.89 \pm 0.042	85.54 \pm 0.15
3	500	53.78 \pm 0.024	90.23 \pm 0.24
4	1000	62.88 \pm 0.016	96.50 \pm 0.16
		IC ₅₀ = 464 $\mu\text{g}/\text{mL}$	IC ₅₀ = 65 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

Table 6: Activity of Methanolic concentrate of *Premna tomentosa* hydroxyl radical method

S.No	Concentrate ($\mu\text{g}/\text{mL}$)	% of activity($\pm\text{SEM}$)*	
		Methanolic concentrate	Ascorbate
1	125	43.66 \pm 0.032	70.34 \pm 0.38
2	250	49.28 \pm 0.046	85.54 \pm 0.15
3	500	67.46 \pm 0.048	90.23 \pm 0.24
4	1000	70.34 \pm 0.056	96.50 \pm 0.16
		IC ₅₀ = 288 $\mu\text{g}/\text{mL}$	IC ₅₀ = 65 $\mu\text{g}/\text{mL}$

* Every value was articulated as mean \pm SEM for 3 experimentation

t omentosa and ascorbic acid were found to be 1038 $\mu\text{g}/\text{mL}$ and 65 $\mu\text{g}/\text{mL}$ correspondingly.

Hydroxyl radical is the most ROS and causes severe injure to adjacent biomolecule. Hydroxyl radical activity of EA concentrate of *Premna tomentosa* and ascorbic acid were showed in Table 5. The hydroxyl radical activity of EA concentrate of *Premna tomentosa* and ascorbic acid at 1000 $\mu\text{g}/\text{mL}$ was found to be 62.88% & 96.50 % correspondingly. The IC₅₀ of EA concentrate of *Premna tomentosa* and ascorbic acid were found to be 464 $\mu\text{g}/\text{mL}$ and 65 $\mu\text{g}/\text{mL}$ correspondingly.

Hydroxyl radical scavenging potency of methanolic concentrate of *Premna tomentosa* and ascorbic acid were showed in Table 6. The hydroxyl radical activity of methanol concentrate of *Premna tomentosa* and ascorbic acid at 1000 $\mu\text{g}/\text{mL}$ was found to be 70.34% & 96.50 % correspondingly. The IC₅₀ of methanol concentrate of *Premna tomentosa* and ascorbic acid were found to be 288 $\mu\text{g}/\text{mL}$ and 65 $\mu\text{g}/\text{mL}$ correspondingly.

Based on the above report the methanolic concentrate of *Premna tomentosa* (IC₅₀ = 288 $\mu\text{g}/\text{mL}$) was found more effective than that of PE(IC₅₀ = 1038 $\mu\text{g}/\text{mL}$) and EA concentrates (IC₅₀ = 464 $\mu\text{g}/\text{mL}$). The methanolic concentrate of *Premna tomentosa* exhibited better hydroxyl radical scavenging activity compared to other concentrates. The scavenging of the hydroxyl radicals possibly may be presence of hydrogen donating ability flavonoids from methano-

lic concentrates.

CONCLUSION

From this investigation, it is concluded that aerial parts of methanolic concentrate of *Premna tomentosa* is established to have valuable potential than PE and EA concentrates. Thus, present data suggest that methanolic concentrate of *Premna tomentosa* is a remarkable resource of natural antioxidant, which might be helpful in preventing the improvement of a variety of free radical induced diseases.

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