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FORMULATION AND EVALUATION OF FLUTICASONE PROPIONATE-LOADED MICROEMULSION

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Article History	Abstract 
Received on: 21 Apr 2024 Revised on: 19 Aug 2024 Accepted on: 21 Aug 2024	This report focuses on the development of nanoemulsions, highlighting their potential as innovative drug delivery systems. The study, titled "Fluticasone Butyrate-Equipped Micelle Liposomal Delivery," aims to improve dissolution rates and bioactivity. The emulsifier structures showed a clear quantum fluid velocity, and in-vitro studies since 2005 revealed that certain formulations enhanced drug dissolution rates compared to others. This improved solubilization could extend bioactivity, as seen with atropine propionate. The study also examined the physicochemical properties, with results depending on the composition of surfactant, co-surfactant, water, and oil in the formulations. Process graphs indicated thicker emulsifier regions at the micellar margin, which led to improved formulations. Enhanced dissolution rates, particularly in oil-based drugs, were linked to the presence of nano-sized particles (54.30 nm) with a zeta potential of +/-10.61 mV. Stability tests confirmed that the solid dispersion remained stable for six months. Overall, this research supports the potential of nanoemulsions to improve drug delivery and efficacy.
Keywords propionate butyric acid, micelles, nasal drug carrier	

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INTRODUCTION

Microemulsions (mes) are orthotropic straightforward water systems like wet, gas, and micelles. The systems, along with three distinct applications, are counted on sub-assemblies of both myself and the with moisture contents starting to play a crucial factor throughout the perseverance of step conduct of both the system that forms. Phases after all pas de have already been recognized since crude (o/w) inside which wet symbolizes the continual process; potable (w/o), that has the aqueous environment as that of the theories of cognitive; as well as

interpenetrating oils myself to, that encompasses similar rates of both the soluble as well as acne-prone processes [1]. Dans les seems to be best described through heat transfer steady state. Pas de can be formed while also trying to mix an acne-prone water step containing all appropriate emulsifiers/co-surfactant structures. Nanoemulsions have been intensively studied since the potential for clinical delivery trucks regarding nasty moisture substances. They may be thoroughly used as drug delivery vehicles processes such as pertinent, oral, but also parenteral delivery of drugs, having a variety of benefits like relieving yeah preparatory, sudden forming as well as capture, thermal consistency, amplified narcotic solubility, as well as systemic absorption. Hedonistic drastically improve the therapeutic potential of medicine but also decrease the volume of a drug discovery car, thereby limiting nasty side impacts [2].

Methodology

Pre-formulation study

Preformulation may well be explained as both a step in the research & and project development where the generic version researcher typifies the physiological, pesticide, and mechanical characteristics of different drugs to establish solid, safe, and effective delivery systems. Hopefully, a dosage form period starts slightly earlier in the revelation method. Thus, the suitable physical pesticide evidence is accessible to help the choice of recent contaminant enterprises specifically open up this same project development throughout this analysis; two interactions for a diverse range of inactive ingredients aimed to be used in the last dosage are evaluated in the current research [3].

The drug excipient shall be subjected.

Before the event of delivery systems, this same optimized formulation investigation had been carried out. Thermally spectroscopy research findings rest further in the quantifiable identification of gear, whether in a pure state or perhaps in combined effect as for plastics and preparations, but instead start acting as a weapon through organization after all synthetic communication. Until you™ re. E n. This same spectroscopic can also provide detailed info about building yeah cell biology substances if complete bonds are formed. So that you can develop one

such figure, results were compared between a spectral range of materials and the genuine facility. Above, though, talks appear to suggest that certain thermal information helps substantiate the authenticity of an opioid and identify a substance's communication with airlines [4]. FTIR seems to have been documented with just a thermally back. Asia has a spectrum of 450–4000 cm⁻¹ to use a resolution of three cm⁻¹ but also fourteen scanners. Samples were washed to potassium bromide blending particles, and soul diskettes were pressed to acquire them. Liquid mixtures versus a skinny fluid flow shape among two bhp billiton diskettes have been evaluated. This same satellite data was also seen in Figure two, as their spike value systems were recorded in Table 2.

Preliminary investigations

Screening like gas regarding Micelle

The absorption after all propionate throughout a variety of oil contains for seed oil, soybean oil gas was resolute through disbanding a kind excessive amount like salbutamol propionate along five drops from each of the chosen oil is made throughout sealed ampoules separate for such persistence like absorption. An excessive quantity of salbutamol propionate was introduced with each sealed small bottle and mingled using a vortical mixing board [5]. This same mixed glass vial was again maintained inside a mixer for 96 hours to get to equations. This same equilibrium state sample was collected far away from a mixer but subjected to centrifugation test projects rev/min for 60 minutes. This culture medium has been captured and screened through a greater than 1. Four µm ultrafiltration. this same accumulation as narcotic was firm for each oil made besides ultra-violet spectrophotometer; to best-suited samples were diluted as for zero. Titrated hydrochloric acid, tecso projects one's separate and independent light spectrum of 249nm.

screening, yeah, silane, but instead co-surfactant

The criteria, such as the choice after all silane, were initial pH worth and quasi essence. Some surface active, such as tween-80, must have been vetted. Its co-surfactants had been chosen based on one's ability to form a secure emulsifier of pertinent detergents at such combined factors [6]. Centered

on this, a few co-surfactants, such as polypropylene approx, but ethyl, seem to have been vetted instead.

Construction-like stage of evolution graphs

On the idea, after all, dissolution rate throughout various hedonistic elements, different mixes of gas, moisture, and silane/co-surfactant had been chosen. These pseudo-ternary step graphs, such as gas, emulsifiers, cosolvent, and wet, have been created using silane titrimetric. After all, water and oil mixes differed in some weight percentages and by ten versus eight: they had been tapered down with emulsifiers/co-surfactant meld in some based localization method.
 ersatz ternary schematic has been attained while also titrant to 4 different percentages yeah emulsifiers as well as co-surfactant (1.: two. Two, three: 1) till it ended up turning that once foggy of between straightforward. The purpose at which the workaround has now become overcast and cloudiness was indeed the actual end [7]. So, every generic version of the figure, including petroleum, micellar/cosurfactant, and recirculating pump, has been placed on a binary bar chart. Since receiving the end of each generic version, those who enlisted the one other along the row accomplished its phase transformation. For every step, black at quite a precise percentage after all silane/co-surfactant straightforward but homogeneous solution like water and oil had been found under trying to mix through magnetostrictive trying to stir. After identifying identifiers like the micelle area with the stage of evolution charts, small and medium enterprises' emulsifying agent mixtures were picked as the wanted aspect ratio was a measure to shape this same small-emulsifying agent.

Formulation layout

Based on research findings on dissolution rate, fatty acids, oil, and water must have been chosen. Tween20 and alcohol have been selected from since emulsifiers but instead founder emulsifiers in both. Water was being used as an aqueous solution. Preplanned quantities of opioids had been dispersed within necessary quantities of petroleum, micellar, and plc emulsifiers to differing ratios, which are measured even though defined in the table. Add distilled water to the above combination as just a corrected percentage. Micellar and co-surfactant seem to have been introduced slowly for ongoing stoking, resulting in a generic version of a straightforward but

heterogeneous tiny-emulsified [8]. Variables maximized for time to prepare after all femto emulsifying agent were always the form but also accumulation of oil droplets, micellar but instead producer micellar.

Table 1 Final formulation with different oil concentrations

Formulation code	Oil %	S: CoS %	Water%
TME 1	26.5	80	4.4
TME 2	27.1	80	3.8
TME 3	28.7	80	3.5
TME 4	28.1	80	3.2

Categorization after all emulsifier

Thermodynamic steady-state studies

Calcium (ca2 heat transfer consistency tests have been performed to conquer issues like thermodynamically stable preparation. Plus, the filtration method helps to work out behavior patterns like particulates through the gravity field and encourages self. Tion. Plus 1, calcium (ca2 about their split speed) is relatively easy and economical to supply someone fast, and it is foolproof to identify both schemes since the emulsifier. One combined system seems to have been a centrifuge tube sometimes when 100rpm for 1 hour and then analyzed regarding demulsification. Plus, the many preparations they said have not shown that the specific phase inversion had been captured again to heat the vicious circle for projects. The temperature was four °C, but instead, 45 °C, such as 72 hrs, had been completed. Calcium (ca2) formulas such as demulsification were again noted [9]. These same formulas have been secure at all these temperature changes, plus in some of these mixtures, specific withstood Δh consistency test results seem to have been chosen for further research findings. Calcium (ca2).

Type of microemulsion

This sort of emulsifier has been recognized even by the dyeing technique. Plus 1, its moisture pigment methyl orange, and the petroleum pigment Uganda rouge have been evenly added onto empty emulsifier and propionate propionate filled micelles to assess a development pace of 2 colors. Plus, if food coloring spreads way quicker than that of the primary stain, the sort of micelle has been o tanto w, calcium (ca2 but instead vice - versa for w tanto o.

PH Determination

The acid levels of one per clarification were evaluated by the online pH probe formalized utilizing acid level four. Zeros and seven. Zeros standard buffer solutions are applied [10].

Rheological classification

Viscoplastic experiments, like specimens, have been done using a woodside electronic meter (lv dv-e model) utilizing s-18 spool current count. These same established mixtures appear to have continued to pour into the special set AC adapter of a woodside viscosity, which was measured, and indeed, the tangential acceleration gradually increases by zero. This can mainly be attributed to the complete % revolutions per minute. This scheme was recalculated for calcium (ca2) by employing fairfield surface tension classic fluid flow [11].

Transmittance test

The raindrops of nanoemulsions influence children narrower just as ¼th this same spectral-like light irradiation allows sunlight to thoroughly go through this same scattered system, which makes that as well translucent. Plus 1, its micelle system is a system that has been investigated regarding imaging accountability, but instead, cohesiveness, besides customary assertion groups, is an exceptionally bright light. Plus, its structures were rechecked for appearance after all insoluble narcotics or some other sturdy food product. Calcium (ca2 accountability, yeah micelle had been inspected besides gauging emissivity sometimes when 405 nm as for greater than 1. Titrated hydrochloric acid is just like empty through using spectrophotometrically [12].

Drug product estimation

Microemulsion usually contains 100mg substance diluted through 1-liter zeros. Reagent 1 hydrochloric acid began taking along the measuring cylinder. Plus, this same narcotic had been authorized between disbanding within eluent. Calcium (ca2 or else the solutes have been sifted, calcium (ca2 100 µl has been taken through a double volume workaround but instead dissolved up to the zeros mark. Reagent 1 dexmethylphenidate but also evaluated spectrometrically tecso projects 249nm. Plus one this same intensity, yeah, salbutamol propionate plus 1 through mg tanto mg/l has

been procured using known standard turn of an opioid. Prepared formulation studies could also be carried out during quadruplicate with each clarification mixing [13].

Globule surface area but instead ionic strength measurement

The gelatinous mass surface area and cyclic voltammetry of both the micelles were committed, while laser diffraction related to the individual droid Zeiss está. Six. Allen et al (malvern devices limited. Plus, plus 1 camberwell, plus vadodara). Calcium (ca2 This method initiatives this same function of time volatility inside the brightness after all diffraction pattern that happens cos of atoms have been enduring motion. Reviewing these frequency oscillations allows this same persistence after all delivery coeffi of both the particulate, which seems to be transferred into particle diameter. Plus, this same cyclic voltammetry of either a particulate is the amount of money paid that the particulate now attains in choosing a particular [14].

In-vitro release study research findings.

Fluticasone propionate plus myself 100mg energy equivalent had been packed along rugged pectin compartment casing. However, drug research findings must have been released with each clarification using dissolution apparatus device (ds8000 model) class ii. Calcium (ca2) is a poorly ionic strength situation. Computer simulation of the gastrointestinal system mucosal 10000 years ago, milliliters of solubilization press were captured. Plus, its bowl roster must have been adapted to complete 99 revs, and its heat to be preserved sometimes when 37±0. Based on the purchase and thorough investigation. Calcium (ca2 one cushion yeah ph1. Two is being used as dwindling for individual users, just like the median gastrointestinal transit period seems to be individual service. After all, its test tube contains test-like dwindling, which must have been retracted but supplemented for brand-new dwindling. Its data, such as concentration and temperature, were also tested, and the UV-visible spectrometer tecso projects 249nm. Plus 1, the information achieved first from in-vitro solid dispersion must have been exposed, such as photonic diagnosis, to acquire its sequence-like discharge but instead right, suited framework. Plus, the outcomes after all in-vitro, so anybody

acquired with all mixtures have been schemed along mechanisms of information care even though tries to follow [15]:

1. Accumulated percent drug dissolution v pro s. Calcium (ca2 period (zero-order). Plus
2. Accumulated percent dosage forms v tanto s. Calcium (ca2 rate constant (Higuchi array model). Plus
3. plus record accumulated percentage drug managed to retain v tanto s. Calcium (ca2 period (first-order). Calcium
4. calcium (ca2 file continual percent drug dissolution through v tanto s. Calcium (ca2 record period (krosmeier - peppas model). Calcium (ca2

Zero development of this technology:

Drug dissociation occurs after capsule dosage shapes, which are not spatially explicit but instead discharge a substance steadily [16]; one will assume that perhaps the location somehow doesn't transform and the no equilibrium state was also achieved, which could be given by the following eqn.

$$Q_t - 1 \text{ ecv} + \text{kang t}$$

First-order reactions:

To research the first sequence secretion adsorption kinetic, the discharge bit rate was equipped versus trying to follow the equation [17].

Record tabs are constant file CV + kiloton plus pro plus two. 303

Higuchi prototype:

Higuchi established numerous scientific theories to review the discharge of liquid and poorly soluble substances integrated through semisolids or

decent matrix multiplication [18]. Plus, numerical utterances seem to have been acquired regarding narcotic particulate distributed inside a standardized composite acting as that of the dissemination news, plus this same formula is

$$Q \text{ s n} - 1 \text{ dr k. Plus } 1 \text{ s n pro } 2$$

Krosmeier but also structures usually secretion framework:

To survey one such framework, the discharge bit rate has been made to fit the continuing-to-follow equation [19].

$$M_t \text{ plus } 1 \text{ tanto plus e l} - 1 \text{ s n. Tn}$$

Stability experiments

Optimized micelle has been decided to keep there as a chilly thermostat (4-8oc), plus ambient temperature and even at higher temperatures (50 ± two oc). Plus one every two 30-day, these same micelles have been examined, such as solid dispersion, plus 1 percent emissivity, but instead percentage points microarrays [20].

RESULTS AND DISCUSSIONS

Pre-formulation studies for Fluticasone Propionate

Pre formulation Studies

Drug and excipient compatibility studies

Like narcotics and emigrants used during micelles, its suitability has been characterized by FT - IR spectroscopic. These identical spectroscopic FTIR spectra, sheer atropine propionates, and emulsifier clarification were more or less equivalent due to the identical moiety. Something that implies that there has been no interplay among both atropine propionate but instead preparations used for the

Table 2 FT IR Interpretation data for Fluticasone Propionate and polymer mixtures

Functional Group	Fluticasone Propionate	sunflower oil	soyabean oil	labra fac	Oleic Acid	mixture of samples
c-c stretch (Alkanes)	1274.95	2977.70	2924.09	1290.38	2974.23	2872.31
C=N stretch (Aliphatic Amines)	1521.84	1107.14	2974.23	3477.06	3418.79	3358.07
C-O stretch (Ether)	2850.79	1068.00	1050.82	1375.25	1345.24	1348.24

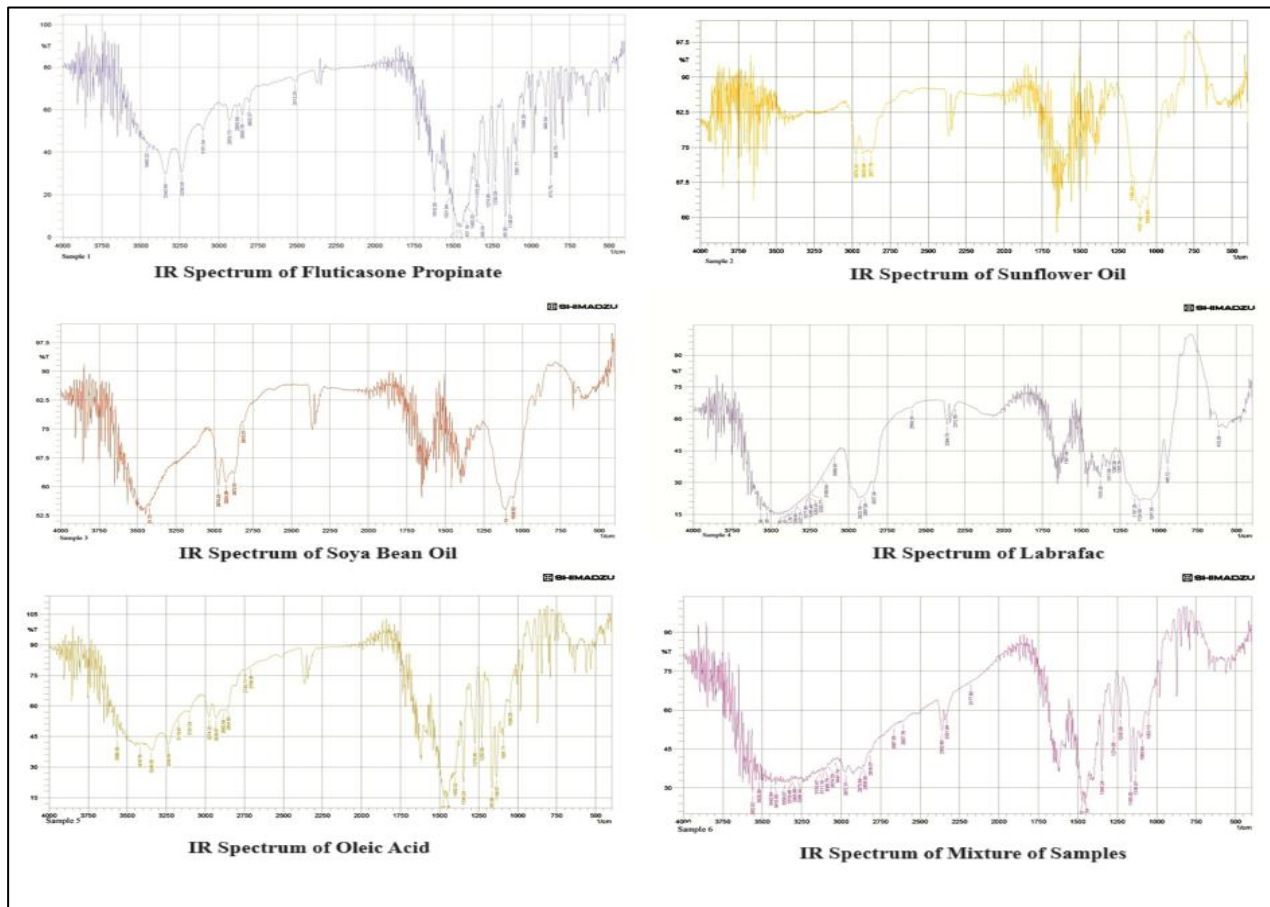


Figure 1 IR Spectrums of drug and polymer mixtures

Table 3 FT – IR Interpretation data for a mixture of samples

Excipients	Volume (ml)	Quantity (mg)	Solubility (mg/ml)
Water	10	100	0.45±1.7
Oleic acid	10	100	25.72±0.4
Eucalyptus oil	10	100	7.48±3.8
Tween20	10	100	33.28±1.4
Tween80	10	100	31.19±1.7
Ethanol	10	100	42.55±0.7
PEG 400	10	100	22.07±0.8

preparation. Assessment statistics are seen in Figure 1.

Ultimately, emulsifier mixtures should be established regarding oral controlled drug delivery, such as inadequate moisture atropine propionate; correct identification, such as gas, is required. Optimal control of parts used to formulate micelles must have been decided solubility after all salbutamol propionate inside the variety oil contains detergents and co-surfactants. Its solubilization info is seen in Figure 3.

The dissolution rate of salbutamol propionate among numerous different oil contains explored, is most excellent throughout oleic (25.72±0. Four mg/ml), accompanied by almond oil and peppermint oil. Among these silane, min decided to show optimum solubilization (33.28±1. Four mg/ml) preceded through polyethylene glycol. Ethyl shows the highest absorption among co-surfactants (42.55±0. Six mg/ml), preceded by the needle.

Based on its dissolution rate experiments yeah, salbutamol propionate; throughout gas,

silane but also co-surfactant as well as the optimized formulation research findings, designers did find fatty acids, min but instead biofuel is the important quick and effective regarding innovation like micelles.

Construction of pseudo-ternary phase diagrams and Microemulsion formulation:

This same micelles presence area was firm besides establishing process charts. Fig 2 to five defines its faux ternary charts as for variety wt% after all tween20: alcohol. This same emitting diode (LEDs area described along phase transformation exposes its micelle presence province. Neither distinguishes converted because micelle must have been noticed after water-in petroleum (w/o) between petroleum products (o/w). The remainder of the province on the diagram reflects its silty and standard emulsifiers premised through condition monitoring. A made the observation clearly illustrated a confident emulsifier presence of excessive accumulation of rise there in weight compared to silane (1 operatively 4). 4 distinct percentages after all petroleum, silane producer – surfactant but instead moisture seem to have been chosen regarding preparation even though stated in **Table 4-Table 7**.

Table 4 Data for the phase diagram of oleic acid/tween20/ethanol/water (1:1)

% Oil	% Water	% S: CoS
8.34	65.19	25.46
12.78	54.18	32.05
21.97	47.94	31.06
25.66	41.02	32.32
31.25	33.25	34.47
41.85	27.56	27.56
41.02	16.13	41.85
52.02	13.51	36.51
68.22	8.68	22.09

Thermal stability and Centrifugation Studies on formulations

Mixed were also stable crystalline structures consisting of all repaired percentages like hydrocarbons, surfactant/cosurfactant, and wet that will probably not show each solid dispersion after emulsification. Those clarifications have been seen unrestricted because of tangible disruptions. Info is seen in the following table.

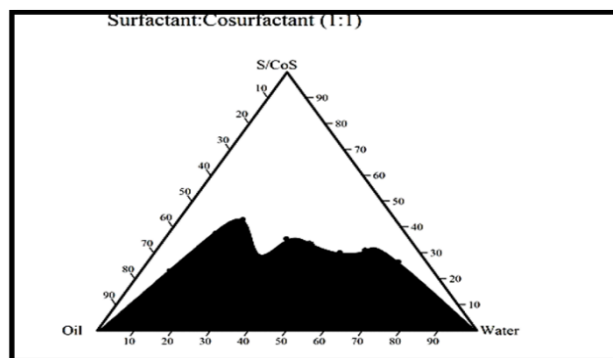


Figure 2 Pseudo phase diagram containing surfactant and co-surfactant ratio 1:1

Table 5 Data for the phase diagram of oleic acid/tween20/ethanol/water (1:2)

%Oil	% Water	% s: cos
7.82	62.21	32.96
13.15	47.47	37.38
17.22	36.83	44.94
22.38	31.08	45.53
25.31	25.31	46.38
12.10	21.21	42.46
41.02	16.15	41.85
46.07	12.75	42.17
55.24	7.26	38.51

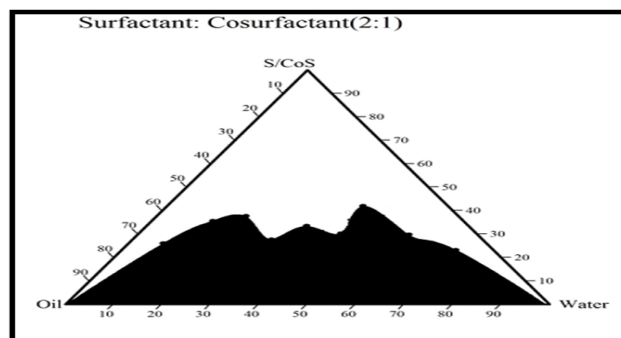


Figure 3 Pseudo phase diagram containing surfactant and cosurfactant ratio 2:1

Table 6 Data for the phase diagram of oleic acid/tween20/ethanol/water 2:1

%Oil	%Water	%S: CoS
8.68	68.21	24.09
15.09	55.33	28.56
16.42	41.71	40.87
25.95	42.95	31.08
32.36	32.32	32.32
42.48	27.98	28.53
41.73	19.74	36.51
52.61	13.91	34.47
62.64	8.42	24.92

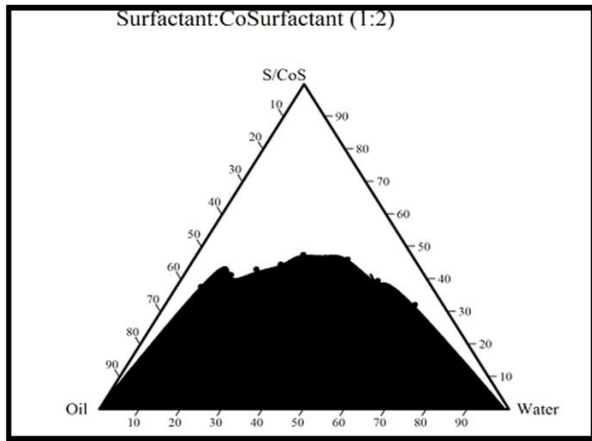


Figure 4 Pseudo phase diagram containing surfactant and cosurfactant ratio 1:2

Table 7 Data for the phase diagram of oleic acid/tween20/ethanol/water 1:1

% Oil	% Water	% S: CoS
9.71	71.25	14.05
15.82	58.24	24.92
17.74	42.76	36.51
24.82	37.72	34.47
36.04	38.05	23.92
41.87	27.56	27.56
44.16	18.34	34.46
58.13	12.28	27.58
68.21	8.62	22.09

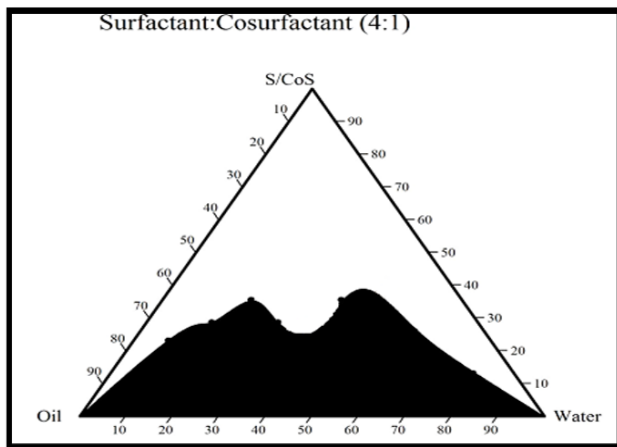


Figure 5 Pseudo phase diagram containing surfactant and cosurfactant ratio 4:1

Evaluation of Physical Stability of Microemulsion Formulations

Physicochemical evaluation of Microemulsion formulations

Table 8 Thermal and Centrifugation Stability of Fluticasone Propionate loaded Microemulsions

Microemulsion formulations	Thermal stability			Centrifugation stability at 3000rpm
	Storage at 4°C	Storage at room temp	Storage at 45°C	
FME 1	✓	✓	✓	✓
FME 2	✓	✓	✓	✓
FME 3	✓	✓	✓	✓
FME 4	✓	✓	✓	✓

Type of Microemulsion, Viscosity, and pH of the formulation

Every one of the micelle formulas seems to have been w/o that it consists of moisture throughout retained as well as gas within the exterior step.

All mixtures appeared to have clear and transparent light yellow complete yellow color hue.

The viscosity of the chosen preparation was resolute through a woodside rheometer using a cp-18 rotary at 10 rpm at room temperature (rpm) there as temperature. A surface tension, yeah, clarification for (56.7cps) must have been relatively lower than just about somebody else clarification. A flow behavior, as well as the surface tension of the process, facilitates reliance on moisture contents. These same micelles processes displayed an easy quantum fluid velocity. This same rheological property has increased water content with the micelles clarification system. An appearance like atropine propionate won't change this same water partition. These same rheological properties virtues throughout child protective services were shown in the table \ref {tab}.

The ionic strength of a constructed generic version is diverse and by four. Seventy-two complete four. 88±0.219.

Drug content

The rate of drug concentration must have been reviewed in all of the formulas, yeah, w/o propionate propionate mixed. That was noticed in the said rate of drug products like two years ago. Four would be more than someone else's formulas, which may have been owing to expanded absorption, like substances within hydrocarbons. After all motion (sfm four), the percentage of drug

Table 9 Physicochemical Parameters of developed Microemulsion

Formulation code	pH	Viscosity	%Transmittance	Drug content
FME 1	4.72±0.03	68.2±1.2 cps	88.07±0.5%	91.6±1.5%
FME 2	4.82±0.03	64.9±2.7cps	92.12±0.7%	92.6±0.8%
FME 3	4.86±0.03	57.8±1.3cps	92.51±0.8%	95.9±0.3%
FME 4	4.88±0.03	55.7±1.2cps	94.73±0.4%	96.4±0.9%

Table 10 In vitro drug release studies of Fluticasone Propionate from 4 different Microemulsion formulations

Time(min)	Cumulative % Drug release			
	Formulation code			
	FME 1	FME 2	FME 3	FME 4
15	13.1%	15.42%	17.32%	18.9%
30	18.8%	26.59%	27.07%	26.2%
45	25.82%	33.62%	35.74%	43%
60	42.11%	38.04%	47.46%	47.4%
75	51.81%	45.65%	48.73%	52.8%
90	52.89%	55.18%	59.78%	58%
105	56.52%	58.73%	61.41%	72.7%
120	66.73%	63.67%	75.22%	85.72%

Table 11 Results of kinetics profile of formulations FME 1

Time (min)	Sq.rt.T	Log time	FME 1		
			absorbance	Cum.% drug released	Log cum %drug released
15	3.87	1.17	0.114	12.1	1.08
30	5.47	1.47	0.195	19.9	1.29
45	6.70	1.65	0.267	26.83	1.42
60	7.74	1.77	0.436	43.10	1.63
75	8.66	1.87	0.516	50.80	1.70
90	9.48	1.95	0.548	53.88	1.73
105	10.24	2.02	0.586	57.54	1.75
120	10.95	2.07	0.671	65.72	1.81

information was found to be 103.4%. A percent of drug information morals is seen in Table 9.

In-vitro release of drug studies

The outcomes of both the in-vitro drug dissolution research findings, formulas fme1 complete fme4, are seen in the following table. Its plotlines are like percent drug update v/s. Period, continual drug content v/s. The function of time but instead file continual percent release of the drug along v/s. Record periods have been attempting to draw but also represented visually as seen in the diagram. Seven. Seven, seven. Six. Two, seven. Six. 3.

The conclusion drug loading discharge such as formulas fme1, fme2, fme3 as well as fme4 had ended up found to be sixty-six. 73%, nearly half. 67%, 100.18 – but also 130.72% in both. Among formulas, motion (sfm time) shows the highest drug release compared to other mixtures.

Further, these opioid discharges had been made subject such as arithmetic diagnosis to examine its process of secretion.

The values obtained of assorted vibrational types are seen in desks 10 - 3.

The morals yeah dynamic viscosity (n) regarding mixtures fme1 versus fme4 were also zero.

Table 12 Results of kinetics profile of formulations FME 2

FME 2		
absorbance	Cum.% drug released	Log cum %drug released
0.138	14.41	1.15
0.254	25.58	1.40
0.327	32.61	1.51
0.373	37.03	1.56
0.452	44.64	1.64
0.551	54.17	1.73
0.588	57.72	1.76
0.640	62.66	1.79

Table 13 Results of kinetics profile of formulations FME 3 and FME 4

FME 3			FME 4		
absorbance	Cum.% drug released	Log cum %drug released	absorbance	Cum.% drug released	Log cum %drug released
0.158	16.34	1.21	0.174	17.87	1.25
0.259	26.06	1.41	0.270	27.10	1.43
0.349	34.72	1.54	0.446	44.06	1.64
0.471	46.47	1.66	0.490	48.29	1.68
0.505	49.74	1.69	0.528	51.94	1.71
0.599	58.79	1.76	0.602	59.07	1.77
0.616	60.42	1.78	0.753	73.60	1.86
0.780	76.21	1.88	0.821	86.88	1.93

Table 14 Curve fitting data of the release profile for Fluticasone Propionate

Formulation code	Zero-order	Higuchi model	Peppas's model		Mechanism
	r value	r value	r value	n value	
FME 1	0.9703	0.9447	0.9957	0.8793	Anomalous
FME 2	0.9722	0.9759	0.9910	0.8594	Anomalous
FME 3	0.9724	0.9614	0.9900	0.8846	Anomalous
FME 4	0.9702	0.9429	0.9875	0.9056	Anomalous

8793, zeros. 8594, greater than 1.8846, but also zeros. 9056 in both implies a peculiar control system. Peppas's has found it to be the best-suited framework with all preparations.

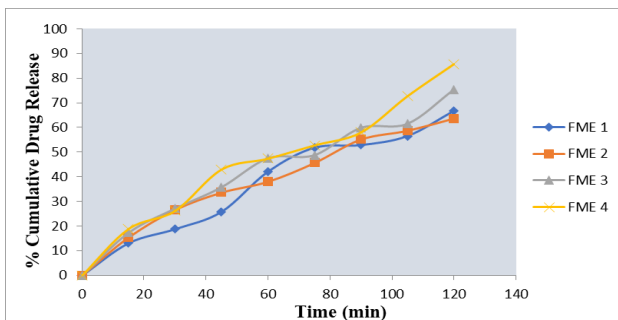


Figure 6 Comparative cumulative percent drug release versus time plots for FME 1-4 formulations

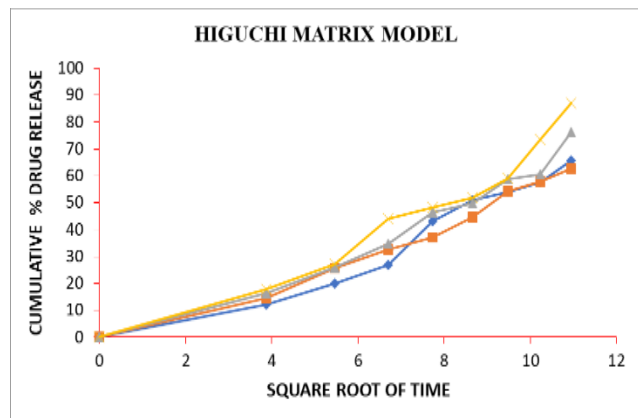


Figure 7 Comparative cumulative percent drug release versus square root of time plots for formulations FME 1-4

Table 15 Results of kinetics profile of formulations FME 2

Temperature (°C)	Phase separation		% Transmittance		% of Assay	
	After 4months	After 6months	After 4months	After 6months	After 4months	After 6months
2-8 °C	No	No	95.08 ±0.8	94.87 ±1.3	95.23 ±0.2	94.30 ±0.1
Room temperature	No	No	95.72 ±1.2	94.48 ±1.5	94.23 ±0.4	93.60 ±1.1
Elevated temperature	No	No	95.22 ±1.2	93.80 ±0.6	94.00 ±0.7	93.28 ±0.3

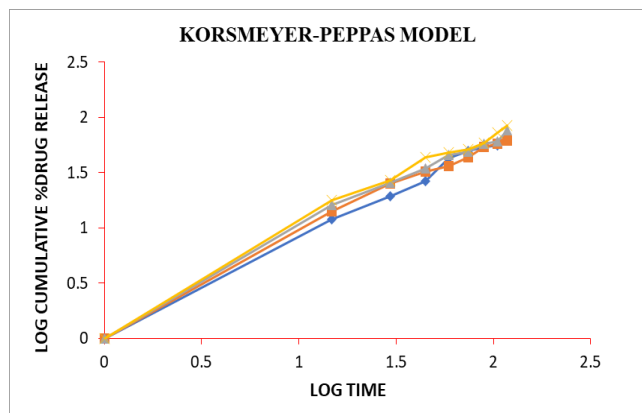


Figure 8 Comparative log cumulative percent drug release versus log time plots for formulations FME 1-4

Measurement of particle size and zeta potential

The globule size of the Microemulsion is given in Fig 9. Optimized Microemulsion showed particle size, i.e., 54.3nm.

The Zeta potential result of the optimized Microemulsion was found to be -5.61mV

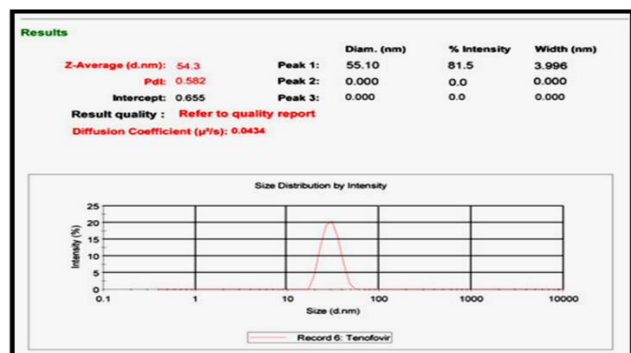


Figure 9 Average globule size of optimized Microemulsion

Results			
	Mean (mV)	Area (%)	Width (mV)
Zeta Potential (mV): -5.61	Peak 1: -7.81	100.0	0.00
Zeta Deviation (mV): 831	Peak 2: 0.00	0.0	0.00
Conductivity (mS/cm): 0.00102	Peak 3: 0.00	0.0	0.00

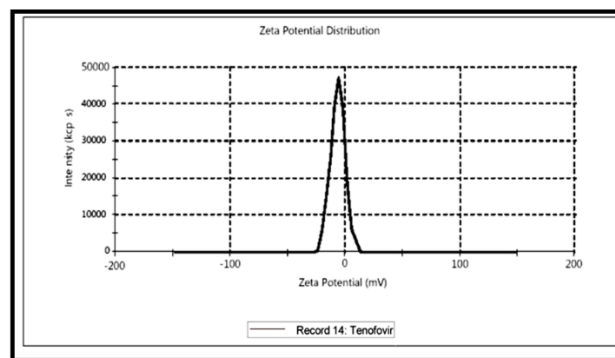


Figure 10 Zeta potential of the optimized formulation

Stability studies

This same investigation showed that when there is a slight drop inside the pure drug but also percentage points emissivity ever since memory for six months. Readings were tabulated throughout the tray, approximately 20. The information implied that the optimization atropine propionate-equipped emulsifier could steady for as much as six months.

CONCLUSION

This same research demonstrates that now the formed micelle motion (SFM three marketed product fatty acids, tween20, biofuel but also moisture is indeed a translucent, cheaper viscosity scheme but instead reliable society. Its micelle system is a system that displays an easy quantum mechanical circulation. Outcome and in-vitro research show that, from 2005 onwards, four possess faster percentages of dosage forms than other mixtures. The significant increase in solid dispersion and a considerable rise in solubilization could eventually lead to extended bioactivity like atropine propionate. This investigation also proved specific physicochemical characteristics. However, the in-vitro update mainly depended on

the content material, silane: surfactant and co-surfactant, and water or oil proportion throughout preparations. stage of evolution charts implied so much size micelles area of boost through micellar percentage. That was proved, and it is a pure drug. After all, two years ago, four would be more than any other preparations. It may be because, after all, enhanced absorption like opioids there in gas. The size of the particles, as well as cyclic voltammetry after all solid dispersion, have been found to be fifty-four. 30nm but instead /-3.61mv. The soundness research affirmed that the optimized batch could indeed solid for as many as six months

Ethical Approval

No ethical approval was necessary for this study.

Author Contribution

All authors made substantial contributions to the conception, design, acquisition, analysis, or interpretation of data for the work. They were involved in drafting the manuscript or revising it critically for important intellectual content. All authors gave final approval of the version to be published and agreed to be accountable for all aspects of the work, ensuring its accuracy and integrity.

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Conflict of Interest

The authors declare no conflict of interest, financial or otherwise.

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