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Development and validation of stability indicating liquid chromatography tandem mass spectrometry method for the determination of Dasatinib in bulk

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ABSTRACT

Dasatinib is a medication used primarily to treat chronic myelogenous leukemia and selected cases of acute lymphoblastic leukemia. In the present study a validated stability-indicating LC-MS/MS method was developed for the estimation of Dasatinib and its degradants on Hypersil ODS (150mm x 4.6 mm i.d., 5μ) column using Methanol: 10mM Ammonium Formate as the mobile phase in a isocratic elution at a flow rate of 1.0 mL/min. The method was validated for accuracy, precision and linearity as per the ICH guidelines. The forced degradation studies were carried out for Dasatinib under acidic, basic, thermal, reduction and oxidation conditions, among which oxidative degradation and basic degradation had above 90 % and 30 % degradation. The degradation products were characterized by MS-MS. The developed method was accurate, precise, specific and rapid found to be suitable for the quantitative analysis of the drug. The recoveries of Dasatinib were found to be within the range.

Keywords: Dasatinib; LC-MS/MS; force degradation; ICH.

INTRODUCTION

Dasatinib is a medication given orally for treating chronic myeloid leukemia and acute lymphoblastic leukemia. It is classified as a kinase inhibitor, which prevent the growth of tumors by reducing the action of proteins that control cell division, growth, and survival. The reduction of these proteins, growth and survival of cancer cells are inhibited. The chemical name for Dasatinib is N-(2-chloro-6-methylphenyl)-2-[[6-[4-(2-hydroxyethyl)piperazin-1-yl]-2-methylpyrimidin-4-yl]amino]-1,3-thia-zole-5-carboxamide (Figure 1). The molecular formula is $C_{22}H_{26}CIN_7O_2S$, where anhydrous free base has a molecular weight of 488.01. The monohydrate corresponds to a formula weight of 506.02.

Literature survey revealed that few bio-analytical methods are developed for estimation ofDasatinib in formulation (Antonio D'Avolio, 2012; Eva Kralj, 2012; Haouala. A, Zanolari, 2009; Hassan Karimi-Maleh, 2016; Irina Andriamanana, 2013; MarekDziadosz, 2012; Michael T. Furlong, 2012; Mohammed G. Kassem, 2013; Silvia De Francia, 2009;StéphaneBouchet, 2011) and stability indicating method of Dasatinibwas reported HPTLC

* Corresponding Author Email: thammababu@gmail.com Contact: +91-8667636779 Received on: 11.09.2017 Revised on: 14.11.2017 Accepted on: 23.11.2017 (Mhaske. D. V, 2007). The present paper presents an effective separation, selective and high sensitive stability indicating method for the determination of Dasatinib and its degradants.



Figure 1: Structure of Dasatinib

MATERIALS AND METHODS

Equipment

High Performance Liquid Chromatography (Shimadzu gradient HPLC system) equipped with a solvent delivery system (Model-LC-10 AT-VP), Rheodyne injector (Model-7725i with 20µl loop). The data were recorded using Class VP data station software. Hypersil ODS (150mm x 4.6 mm i.d., 5µ) column was used for method development and validation at ambient temperature (25°C). In this current method the separation was

achieved using 1ml/minflow rate, detection at 290nm with an injection volume of 20µl. LC-MS/MS (Shimatzu 8030, Tokyo japan) equipped with electrospray ionization (ESI) interface, LC-20AD pump, SPD-M20 PDA detector, CTO-20AC column oven, CBM-20 alite controller and SIL-20AC auto sampler using Lab solution data station.

Forced degradation

Stress studies were carried out according to ICH guidelines Q1A(R2). The API was subjected to forced degradation under acidic, basic and neutral conditions by refluxing in 1 N HCl, 1N NaOH at room temperature and oxidative stress studies was carried out using 30% H₂O₂ at room temperature for 2, 4, 6, 8, 12 and 24hr, respectively. at 350°C, CID gas maintained at 230Kpa, N_2 was used as nebulizing and drying gas flow was set at 3 L/min and 15 L/min respectively.

Validation parameters

A stock solution containing 1 mg/ mL was diluted to produce a concentration range of 10–200 μ g/mL to establish linearity and range. The analysis was carried out in triplicate by injecting. The intra- and inter-day precision were established by analyzing for QC concentration of 50, 100 and 200 μ g/mL of drug solution (Table 1), respectively. Accuracy of the method was determined by analyzing QC concentrations 50, 100 and 200 μ g/mL spiked with stressed sample in triplicate and then determining the percent recovery (Table 1). The signal to noise ratios were 3:1 and 10:1 for determining limit of

Actual Conc. (μg/ml)	Recovered Conc (μg/ml) ±SD; RSD% (n=6)	% Recovery	Intraday calculated concentration (μg/mL) ± S.D.; RSD % (n=5)	Interday calculated concentra- tion(µg/mL) ± S.D.; RSD % (n=5)	
50	49.92 ± 0.01, 0.021	101.58	50.00 ± 0.11, 0.222	50.00 ± 0.19, 0.375	
100	99.81 ± 0.17, 0.342	101.61	100.00 ± 0.25, 0.250	100.00 ± 0.21, 0.231	
200	199.32 ± 0.15, 0.152	98.32	199.64 ± 0.27, 0.314	199.99 ± 0.26, 0.129	
125 100 0.55 0.00 100 100 150		45 10 525 10 567 10 50 550 m2		B B B B B B B C B C B C C B C C B C	
C	2.261 2.00 4.22 4.22				

Table 1: Recovery, Precision studies for the Dasatinib

Figure 2: Dasatinib standard mass scan spectra (A), standard LC chromatogram (B), degradation peak of basic hydrolysis (C) and oxidative degradation peak (D)

Chromatographic condition

For the separation Hypersil ODS (150mm x 4.6 mm i.d., 5μ) column using Methanol: 10mM Ammonium Formate as the mobile phase in a isocratic elution at a flow rate of 1.0 mL/min. The column effluents were monitored by a UV detector set at 325 nm. A volume of 20 μ L was injected into the system at a run time of 10 min. The operation condition for MS/MS was The Mass parameters were as follows: Probe temperature was ambient, DL temperature was 250°C, Block temperature was set

detection and quantitation, respectively.

RESULTS AND DISCUSSION

Validation

The method was validated over the concentration range of 10–200 μ g/mL (r2 = 0.9992) of the drug. The data analysis showed that the % RSD for each concentration was <0.25%. The % RSD for intra- and inter-day precision at three different concentrations, viz., 50, 100, and 200 μ g/mL was found to be within the limits (<0.40%) as

S.No	Time (hrs)	Basic hydrolysis (% degradation) 0.1N NaOH	Acid hydrolysis (% degradation) 0.1N HCl	Neutral degradation (% degradation) H ₂ 0	Oxidative degradation (% degradation) 3% H ₂ O ₂	Photo degradation (% degradation) UV
1	0	0	0	0	0	0
2	2	1.73	1.58	1.38	3.87	1.14
3	4	5.82	3.26	3.15	18.74	2.36
4	6	13.29	4.82	5.62	35.53	3.74
5	8	22.67	6.86	7.44	60.26	5.21
6	12	29.14	8.49	8.07	75.14	6.10
7	24	33.36	10.52	10.58	94.52	8.63

Table 2: Degradant observed for Dasatinib from various stress conditions



Figure 3: Mass spectra of the degradant NB-I (A) and NB-II (B)

shown in Table 1. Further, stressed samples with spiked known concentration of the drug, produced good recoveries with mean recovery of 100.50% (Table 1). The limits of detection (LOD) and quantification (LOQ) were found to be 10 and 30 ng/mL, respectively. The capacity (k1) and resolution factors (Rs) were found to be within the limits.

Degradation

The degradation behavior of Dasatinib was studied under various stress conditions such as Acidic, Basic, Neutral, Oxidative and photo-degradation as shown in Figure 2. The percentage of the degradant observed from various stress conditions are shown in Table 2.

Acid, Base and Neutral Hydrolysis

Hydrolysis was carried out by diluting a known amount of stock solution with 1N HCl (acid), 1N NaOH (base) and H₂O (neutral) to produce a concentration of 50 μ g/mL. Finally the samples were analyzed at different time intervals upto 24 hr and the degradants were recorded and collected.

Oxidative degradation

Oxidative degradation was performed by diluting the

stock solution with 30% H_2O_2 to produce a concentration of 50 $\mu g/mL$, which was analyzed at different time intervals.

Photolytic degradation

For photolytic degradation studies stock solution was diluted with water to produce a concentration similar to other studies that is 50 μ g/mL. Further the drug substance in solid state was exposed to a temperature of 80°C for 24 hr, which was diluted to produce a concentration of 50 μ g/mL and analyzed.

Mass spectral fragmentation

Among the degradations peak produced the degradation peak NB-I and NB-II were the major degradation peak observed in oxidative degradation and basic hydrolysis (Figure 3). For the identification of the molecular ion the separation was done by column (HPLC) and the samples were directly injected into a MS/MS system which consists of a the mobile phase water and acetonitrile in the ratio 20: 80, v/v with the flow rate of 0.5 mL/min, using nitrogen as the drying gas, nebulizer gas, 60 psi; dry temperature, 350°C; capillary voltage, 5kV; vaporizer temperature, 400°C; and dwell time, 200 ms.

CONCLUSION

A stability indicating method has been developed to explore the degradation behavior of Dasatinib by an LC-MS/MS. The method developed was able to separate the degradants from Dasatinib The degradation products formed upon basic hydrolysis (NB-I) may be 2-[(6-amino-2-methylpyrimidin-4-yl)amino]-N-(2-chlorocy-clohexyl)-1,3-thiazole-5-carboxamide and degradant

formed during oxidative hydrolysis might be 2-[(6amino-2-methylpyrimidin-4-yl)amino]-N-(2-chlorocyclohex-3-en-1-yl)-1,3-thiazole-5-carboxamide. Hence the developed method can be used for quality control test and identification of degradation product.

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