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Comparative study of atenolol in human plasma by high performance liquid chromatography and capillary zone electrophoresis

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INTRODUCTION

Atenolol is a beta-blocker medication that is often used to control high blood pressure and heart disease (specifically chest pain due to heart disease) (Atenolol Monograph for Professionals, 2018). 4-(2-hydroxy-3-isopropylaminopropoxy) phenylacetamide, also known as atenolol, is a beta-adrenergic receptor blocking agent used in the [Indian Pharmacopoeia \[Indian pharma](#page-4-0)[copoe](#page-4-0)ia;1996]. While atenolol is quickly absorbed, just about half of it reaches the gastrointestinal tract. During oral administration, the average amount of atenolol absorbed is around half of the

total atenolol concentration. The rest is excreted in its natural state in the urine. The kidneys are the main organs from which atenolol is removed (Wander *et al.*, 2009; Heel *et al.*, 1979). Atenolol has antihypertensive effects, which are due to its blood pressure-lowering effect, which is achieved with a single daily dosage (Miller, 1991). A peak pl[asma](#page-5-0) [concentra](#page-5-0)t[ion o](#page-5-0)f [200-300 n](#page-5-1)g[/ml i](#page-5-1)s obtained as a result of this procedure (Braza *et al.*, 2000).

HPLC (Chiu *et al.*, 1997; Chatterjee *et al.*, 1995; Morris *et al.*, 1991), L[iquid chroma](#page-5-2)tography-tandem mass spectrometry (Li *et al.*, 2007; [Gonza](#page-4-1)lez *et al.*, 2010), and GC-MS (Yilm[az and Arsla](#page-4-1)n, 2009; Amendola *et al.*, [2000\) a](#page-4-2)r[e som](#page-4-2)[e of the methods](#page-4-3) u[sed to](#page-4-3) [determine atenolo](#page-5-3)l.

The measurement o[f atenolol in h](#page-5-4)[uman plasma is](#page-5-5) [gener](#page-5-5)ally d[one u](#page-4-4)sing fluorometric technique[s \(Dale](#page-4-4) [and Turne](#page-4-4)r, 1990; Miller, 1991). Many of these methods (De Abreu *et al.*, 2003; Hebert *et al.*, 2005) depend on time-consuming multi-step liquid or solid-phase extraction procedures. As a result, [mini](#page-5-7)[mum sample volum](#page-5-7)[e requi](#page-5-2)r[ement](#page-5-2)s have been established (Sp[anakis and Niopas,](#page-5-8) 2013; [Gwak and Chun](#page-5-9), 2007). Figure 1 shows the structural equations for atenolol.

Figure 1: Structure of Atenolol

The overall goal of this research is to develop and validate a straightforward, selective, and repeatable HPLC mechanism. Capillary electrophoresis is being used to demonstrate the procedure's suitability for this function. HPLC and CZE are used to determine the amount of atenolol in human plasma using Ses techniques.

MATERIALS AND METHODS

HPLC and Capillary Zone Electrophoresis

Apparatus

A Waters Alliance HPLC 2695, which includes a pump, autosampler, and fluorescence detector, was used for chromatography. Using the empower chromatography manager method, data was obtained on a computer.

A Hewlett Packard HP 3D CE capillary electrophoresis device with a diode array detector was used to carry out CZE.

Chemicals and Reagents

Both reagents were of the highest analytical reagent grade. Atenolol and nadolol are acquired from a pharmaceutical producer. These products were all of HPLC quality: Acetonitrile, methanol, 85% phosphoric acid, sodium hydroxide, sodium lauryl sulphate, and sulfuric acid were all HPLC grade. Hollow-polymer ion-exchange (HP-HILIC) grade water was prepared using reverse osmosis and was further filtered using a Milli-Q method.

In the CZE, Merck provides analytically pure Methanol and n-hexane; the majority of the chemicals were supplied by Merck.

Chromatographic Conditions

A Symmetry shield RP¹⁸ 4.5*×*150mm column, guarded by Pak pre-column module with Nova-Pak $C₁₈$, was used. Mobile phase containing 10mM Sodium hydrogen phosphate (ph=3) mixed with 7.3mM sodium lauryl sulphate, methanol and acetonitrile $(40:57:3, v:v:v)$. And it was filtered through a $0.45 \mu m$ size and delivered at a flow rate of 1.0 ml/min. A fluorescence detector is used and its wavelength of 229 and 289nm.

Electrophorectic Conditions

The fused silica capillaries had an I.D. of 58.5mm and an O.D. of 75*µ*m with a detection window of 50mm. This allowed for the detection wavelength to be 194nm. The capillary temperature was set at 25.0*◦*C *±* 0.1*◦*C. Buffer was formed by mixing 50mM sodium tetraborate and 50mM boric acid at a ratio of 50:50 (v/v), which produced a pH of 9. Although the concentration of the salt and acid is 0.5M, the buffer was used to carry out the clean-up process, which allows the buffer to be polluted.

Capillary Conditioning

Capillary was primed on a daily basis with an initial wash cycle consisting of 15 minutes of 1M NaOH treatment, 10 minutes of deionized water, and 5 minutes of electrolyte run. The capillary was rinsed at the end of each injection. The capillary was rinsed with 1M NaOH for 10 minutes before being rinsed with deionized water for 10 minutes in the regular experiment. It was then flushed for three minutes of fresh air.

Experiments

High-Performance Liquid Chromatography

Preparation of Standard and Quality Control Samples

A stock solution of Atenolol and IS were prepared in Methanol and diluted with blank human plasma to produce a working solution of 10*µ*g/ml and 2.0*µ*g/ml. Nine calibration standards in the range of 0.01 -15 μ g/ml and 4 quality control samples(0.01,0.03,0.8 & 1.4*µ*g/ml) were prepared in human plasma. Samples are vortexed for 1min & transferred to glass culture tubes and stored at -20*^o* C until used.

Sample Preparation

In each tube, 100μ of IS solution & 100μ of 0.1M NaOH was added and vortexed for 10sec. After that, with the addition of 3.5ml of solvent, again the samples are vortexed for 1min and centrifuged at 4000rpm for 1min. The organic layer was transferred into a clean tube containing 0.05M Sulfuric acid. 150μ l of the aqueous layer was transferred into an autosampler & 50*µ*l were injected into an HPLC system with a run time of 10mins (Yusuf *et al.*, 2016).

Method Validation

The method was validated according [to USFDA](#page-5-10) [guide](#page-5-10)lines: The parameters include accuracy, precision, linearity, recovery and stability (FDA, 2018).

Capillary Zone Electrophoresis

Stock Solution

Because of its tolerance to phot[odegradati](#page-5-11)on, atenolol was made in water and stored in an amber glass volumetric flask. To allow for further dilution, a solvent is prepared in advance.

Sample Collection and Sample Clean Up

Blood was extracted by syringe and was transferred into tubes containing EDTA. Performed at 5000 RPM, combined, and centrifuged. The next move was to allow the plasma to thaw at room temperature and then add 1ml of human plasma. In this experiment, vortex mixing was done and then 20*µ*l of perchloric acid was applied to precipitate proteins. After vortexing the mixture once, it was centrifuged for 5 minutes at 5000 rpm to isolate the solids and change the pH to 9.

In order to neutralize and activate the Bond-Elut Certify LRC cartridge, 2ml of plasma solution was used. This solution was previously combined with 4ml of methanol and deionized water. When the samples have been placed in the column, they are pulled into the column by gravity. Then, the cartridge was put in a dish, which held a 2ml mixture of water, 1ml of 1M Acetate buffer, and 1ml of n-hexane. The atenolol was derived with 2ml of chloroform-isopropanol (80:20, v/v). The removed components were evaporated to dryness at 50*^o*C under nitrogen and then the aqueous solution was reconstituted with 50*µ*l of deionized water (Arias *et al.*, 2001).

Quantitative Parameters

The spike human plasma samples were infused with known quantities of atenolol, as well as samples containing unknown concentrations of atenolol. With the peak area of atenolol vs the concentration of atenolol, the calibration equation was discovered using least-squares regression. The detection and quantification limits were calculated to be detectable and quantifiable (Meissner *et al.*, 1998; Minamide *et al.*, 2011).

The precision was estimated at two concentration levels, with spiked plasma being matched to c[oncen](#page-5-12)[tration and then prod](#page-5-13)ucing a [calibration map](#page-5-12).

The recovery was noticed by calculating the peak area of atenolol.

A measured concentration of three was chosen for the reproducibility test: 100, 200, and 400ng/ml. In order to make assumptions about intraday and interday activity, Figure 2

Figure 2: Time of injection and height of the peak of atenelol

RESULTS AND DISCUSSION

Optimization of Chromatography Conditions

The mobile process is used in the tests and is distributed at a 1ml/min flow rate. Within 10 minutes of running time, the atenolol, IS, and plasma components separated under these conditions. As a result, the retention times of atenolol and IS were discovered to be 5.4 and 8.3 minutes, respectively, as seen in Figure 3.

Figure 3: Retention time for Atenolol and IS

Linearity was evaluated by analyzing 8 curves, each with nine standard concentration is shown

| Nominal level $(\mu$ g/ml) | Measured level $(\mu g/ml)$ Mean SD | | CV(%) | Bias $(\%)$ |
|-------------------------------|--|------|-------|-------------|
| 0.01 | 0.01 | 0.00 | 13.9 | 15.9 |
| 0.02 | 0.02 | 0.00 | 8.6 | 0.2 |
| 0.05 | 0.05 | 0.00 | 2.4 | -1.3 |
| 0.10 | 0.10 | 0.00 | 2.6 | 0.3 |
| 0.25 | 0.25 | 0.00 | 1.6 | 0.9 |
| 0.50 | 0.50 | 0.01 | 1.1 | 0.0 |
| 0.75 | 0.75 | 0.01 | 1.5 | -0.4 |
| 1.00 | 1.00 | 0.02 | 2.1 | -0.3 |

Table 1: Back-calculated Atenolol concentration from eight calibration curves

The limit of quantification, defined as the lowest measured concentration with precision and bias of \leq 20%, was 10 μ g/ml.

Table 3: Recovery of Atenolol & IS from 0.23ml human plasma

Table 4: Variations in the parameters of analysis of atenelol

Figure 4: overall chromatograms

Figure 5: The Retention time for Atenolol

in Table 1. The calibration curve were linear R²≥0.9992. Mean concentrations back-calculated from peak height ratios using individual regression equations and overall chromatograms are given Figure 4.

Precision and BIAS

To reliably forecast intra-day and inter-day accuracy by [usi](#page-4-6)ng the amounts of atenolol at various intervals. Over the timeframe, from 2.1% to 5.3% and from 0.0% to 17.0%, it varied from 2.1% to 5.3% and from 0.0% to 17.0%. The inter-day varied from 3.1% to 5.2%. Table 2 outlines the conclusions.

Recovery

For direct comparison of peak height from plasma and mobile [sa](#page-3-0)mples, using 5 replicates for five QC samples, 5 cm replicates were employed. In comparison, the recovery of the Islamic State remains unchanged. The values are listed in Table 3.

Robustness and Ruggedness

Robustness is characterized as a slight shift in chromatographic conditions. to assess the rob[us](#page-3-1)tness of the current assay, we attempted to alter the power of the buffer. Theoretically, no significance effects were determined. Ruggedness was put to the test by doing split sample testing. Two different kinds of samples were sampled by two different types of analytical equipment.

Quantitative Determination of Parameters in CZE

The calibration curve for atenolol in plasma is created by generating a calibration curve using spiked plasma blanks and then extracting the samples with the SPE process. The electrophoretic separation takes place in less than three minutes. And, the graph can be seen in Figure 5. The Parameters values are shown in Table 4.

CONCLUSION

The HPLC and CZE as[sa](#page-3-2)y method is precise, simple and effective. In HPLC, the Liquid-Liquid Extraction method is used, wherein the CZE Solid Phase Extraction method is used. The assay was applied to monitor atenolol under various conditions. CZE, the electrophoretic separation takes place in less than 3mins, wherein HPLC the retention time is within 10mins. So, CZE has a better separation than HPLC.

Conϐlict of Interest

The authors declare that they have no conflict of interest for this study.

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