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Development and Validation of a Headspace Gas Chromatographic (HS-GC) Method for Determination of Residual Solvents in Nitaz[oxanide](www.ijrps.com) API

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INTRODUCTION

Several organic volatile solvents or chemicals are used in the manufacturing of drug substances, excipients and drug products. They are also used to increase the final yield, to enhance the purity or to change the physical form such as polymorphic form and solubility. These solvents or chemicals do not have any therapeutic activity but maybe toxic for humans if consumed more than permitted daily exposure (PDE) (Sitaramaraju *et al.*, 2008). It is necessary to remove them, but some solvents remain in small quantities in the final product. These small quantities of organic solvents remain in the final product is know[n as residual solvents. D](#page-8-0)etermination of these residual solvents from drug substances,

excipients and drug products is a difficult and challenging task. Headspace gas chromatographic technique is most suitable and used for the determination of residual solvents. The acceptance limit for residual solvents is set following the toxicity of solvents and specified in the international conference on harmonisation Q3C guidelines (ICH, 2016; Harold *et al.*, 1997).

Nitazoxanide (Figure 1) is chemically 2-acetyloxy-N-(5-nitro-2-thiazolyl) benzamide and have broad[spectrum antipro](#page-8-2)tozoal and antipara[sitic activ](#page-8-1)ity (Rossignol and Cavier, 1976). It is also used to treat helminthic, [p](#page-3-0)rotozoal, and viral infections. Cryptosporidiosis and giardiasis in immunocompetent patients also be treated with nitazoxanid[e \(Rossignol](#page-8-3) *et al.*, 2001; [Még](#page-8-3)raud *et al.*, 1998). It is a prodrug and absorbs from the gastrointestinal tract when administered orally. In humans, it rapidly hydrolysed to its active metabolite tizoxanide [\(Korba](#page-8-4) *et al.*, 20[08\).](#page-8-4)

Literature survey revealed that several methods by UV – spectroscopy (Pandey, 2009; Gandhi *et al.*, [2008\), visible spe](#page-8-5)ctroscopy (Narayana and Manohara, 2007) and liquid chromatography (Malesuik *et al.*, 2009; Kumar *et al.*, 2009; Narayan and Mahendra, 2007) are re[ported](#page-8-6) f[or qua](#page-8-6)n[titative](#page-8-7) [estim](#page-8-7)a[tion \(](#page-8-7)assay) of nitazoxanide in [bulk alone and](#page-8-8) [combination wit](#page-8-8)h other drugs. During the syn[thesis](#page-8-9) and purific[ation](#page-8-9) [process of nitazoxan](#page-8-10)[ide API, Ace](#page-8-11)[tone, Dichl](#page-8-11)o[romet](#page-8-11)hane and Cyclohexane were used. This work aimed to develop and validate a simple and sensitive headspace gas chromatographic (HS-GC) method for simultaneous determination of Acetone, Dichloromethane and Cyclohexane in Nitazoxanide API. Acetone, Dichloromethane and Cyclohexane belong to class 3, 2 and 2 respectively. The specifications as per international conference on harmonisation Q3C guidelines are tabulated in Table 1.

MATERIALS AND CHEMICALS

GC reference standards of Acetone, Dichloromethane (Methylene dichloride) and cyclohexane were procured from Biosolve Chimie, France. Dimethyl formamide (N, N-Dimethyl Formamide) (Supra Solv, GC grade) was procured from Merck Millipore, India. Nitazoxanide API sample was received as gift samples from Suven Life Sciences Limited, Hyderabad, India.

METHOD

The method was developed and validated on Agilent Technologies gas chromatograph (Model

No. 7890B) and a headspace sampler (Model No. 7697A) equipped with flame ionisation detector (FID) using Empower 3 software. The separation of analytes was achieved with DB – 624 (30 m length, 0.53 mm inner diameter and 3.0 *µ*m in film thickness) capillary column. The chromatographic parameters were optimised, and optimised chromatographic conditions are shown in Table 2.

Diluent

Dimethylformamide.

Blank

Use diluent as blank.

Preparation of Standard solution

Weigh accurately about 500 mg, 60 mg, 388 mg of Acetone, Dichloromethane and Cyclohexane reference standards respectively into a 100 mL volumetric ϐlask having about 25 mL of diluent. Mix and make up to volume with diluent. Transfer 5.0 mL of above solution into a 100 mL volumetric flask and dilute to volume with diluent and mix well. Transfer 2.0 ml of the above solution into six different headspace vials and seal properly.

Preparation of Sample solution

Weigh and transfer accurately about 100 mg of Nitazoxanide API sample into a headspace vial. Add 2.0 mL of diluent, dissolve and seal the vial properly.

Preparation of System Suitability solution

Use the standard solution to check the system suitability.

Procedure

Inject blank (1 injection), and standard solution (6 injections), sample solution (1 injection) into the chromatograph and record the peak response of eluting peaks using the chromatographic and Headspace parameters.

Acceptance criteria for System Suitability

The resolution between Acetone and Dichloromethane peaks from the first standard injection from system suitability should be not less than 3.0.

The relative standard deviation (RSD) of area response for Acetone, Dichloromethane and Cyclohexane peaks between the six replicate injections of the standard should be no more than 10 %.

Method validation

Validation of the developed method was conducted as per United States Pharmacopoeia general chapter <1225> (USP, 2018a) and International Conference on Harmonization Q2 (R1) (ICH, 2005) guidelines.

System suitability

System suitability was evaluated under United States Pharmacopoeia general chapter <621> (USP, 2018b). System suitability of the method was established by injecting blank and standard solution for system suitability, calculated the resolution between Acetone and Dichloromethane [peaks](#page-8-13) from first standard injection from system suitability and the relative standard deviation (RSD) of area response for Acetone, Dichloromethane and Cyclohexane peaks from the six replicate injections of the standard solution. The acceptance criteria for resolution between Acetone and Dichloromethane peaks was not less than 3.0 and % RSD for area response of Acetone, Dichloromethane and Cyclohexane peaks were not more than ten from six replicate injections of the standard solution.

Specificity

The specificity of the method was established by injecting blank in triplicate, standard solution, test solution, test solution spiked with analytes at the specification level, Acetone reference standard solution at the specification level, Dichloromethane reference standard solution at specification level and Cyclohexane reference standard solution at the specification level. The chromatograms were evaluated for any interference at the retention time of Acetone, Dichloromethane and Cyclohexane peaks.

Precision

The precision of the method was evaluated by injecting six test sample preparations spiked with Acetone, Dichloromethane and Cyclohexane reference standards at 100% specification level. % relative standard deviation of six test sample preparations spiked with analytes was calculated. Intermediate precision of the method was also evaluated using different analyst, different day, different instrument and different column by injecting six test sample preparations spiked with analytes prepared as same for precision. The acceptance criteria for individual precision % RSD was not more than 5.0, and for 12 preparation results was not more than 7.0.

Accuracy (Recovery)

Recovery study was performed to evaluate the accuracy of the method by spiking method. Recovery study was done by spiking Acetone, Dichloromethane, and Cyclohexane reference standards into the test sample in the concentration of LOQ, 50%, 100% and 150% level of the proposed specification concentration. The recovery samples were prepared in triplicate for 50% & 100% level and six preparations for LOQ & 150%. Injected the prepared recovery samples in the optimised

experimental conditions. % recovery of Acetone, Dichloromethane and Cyclohexane peaks were calculated for all the levels. The acceptance criterion for recovery of Acetone, Dichloromethane and Cyclohexane analytes was 80.0 to 120.0% and % RSD for six recovery results at LOQ, and 150% was not more than 5.0.

Limit of detection (LOD) and limit of quantitation (LOQ)

LOD is the lowest amount of analyte that can be detected, but not necessarily quantifiable, and LOQ is the lowest amount of analyte that can be quantitated with acceptable precision and accuracy. The LOD and LOQ were established by injecting a known concentration of serial dilutions of Acetone, Dichloromethane and Cyclohexane under the stated experimental conditions. The LOD and LOQ were established from the Slope and STEYX by plotting the linearity curve of concentration versus area response. LOD and LOQ were estimated by using the following formulae:

$$
\text{LOD} = 3.3 \times (\sigma/S)
$$

 $LOQ = 10 \times (\sigma/S)$

Where σ = STEYX of response and S = slope determined from the linear plot.

Linearity

The linearity of the method was established for Acetone, Dichloromethane and Cyclohexane from LOQ to 150% of the proposed concentration using six calibration levels (LOQ, 25, 50, 100, 125 and 150% of the targeted concentration). The reference standards were used to prepare calibration levels. The calibration curves for Acetone, Dichloromethane and Cyclohexane, were plotted for each level as concentration versus peak area response. The results of linearity were evaluated by regression analysis.

Robustness

Robustness of the method was determined with deliberate changes in the method conditions from the optimised final conditions. Injected blank, standard solution, test sample and spiked test sample solution in each robustness conditions and evaluated the system suitability.

For robustness study, the following parameters were considered such as (i) Change in initial oven temperature 60*◦*C *±*5 *◦*C (55*◦*C and 65*◦*C) and (ii) Change in nitrogen gas flow rate 3.0 mL/min $\pm 10\%$ (2.7 and 3.3 mL/min).

Solution stability

Solution stability was established for the standard solution and test sample preparations. Bench-top

(controlled room temperature) stability was established by injecting standard solution and test sample at regular interval for 48 hours. Solution stability was established by calculating the similarity factor for the standard solution against a new standard and % difference for a test sample from the initial value.

RESULTS

System suitability

System suitability of the method was evaluated through resolution between Acetone and Dichloromethane peaks from standard solution and the % RSD of area response for Acetone, Dichloromethane and Cyclohexane peaks from the six replicate injections of the standard solution. The system suitability results were found well within the predefined acceptance criteria. The results are presented in Table 3.

Figure 1: Chemical structure of nitazoxanide

Figure 2: Chromatogram of blank

Specificity

The specificity of the method was performed to check blank interference and confirm the identity of the analytes. The chromatograms confirm (Figures 2, 3, 4 and 5) no interference at the retention time of Acetone, Dichloromethane and Cyclohexane peaks peak due to blank.

Prec[is](#page-3-1)i[on](#page-3-2)

The preci[sio](#page-3-3)n of [th](#page-3-4)e method was evaluated by injecting six test sample preparations spiked with Ace-

Figure 3: Chromatogram of standard

Figure 4: Chromatogram of the test sample

Figure 5: Chromatogram of a spiked test sample

Figure 6: Chromatogram at LOQ level

Figure 7: Chromatogram at 150 % level

Table 1: Specifications of residual solvents

Table 3: System suitability results

Table 4: Precision results

Table 5: Accuracy results

Table 6: Determination of LOD and LOQ

LOD= Steyx * 3.3 / Slope; LOQ= Steyx * 10 / Slope

Table 7: Linearity

Figure 8: Linearity graph of acetone

Figure 9: Linearity graph of dichloromethane

to be 2.2, 2.6 and 2.3, respectively for Acetone, Dichloromethane and Cyclohexane. The results were found well within the acceptance criteria. The results of precision are presented in Table 4.

Accuracy (Recovery)

The accuracy of the method was evaluated by calculating the recoveries at LOQ, 50%, 1[00](#page-5-0)% and

Figure 10: Linearity graph of cyclohexane

150% level of the targeted specification concentration. The mean % recoveries for Acetone, Dichloromethane and Cyclohexane at LOQ (n=6), 50% (n=3), 100% (n=3), and 150% (n=6) were found within the acceptance criteria. The % RSD at LOQ was found 4.7, 4.6 and 3.7 respectively for Acetone, Dichloromethane and Cyclohexane. At 150% level the % RSD for Acetone, Dichloromethane and Cyclohexane was found 2.5, 2.8 and 2.4 respectively. The recoveries and precision at LOQ and 150% were found within the acceptance criteria. The results of accuracy are presented in Table 5.

Limit of detection (LOD) and limit of quantitation (LOQ)

The LOD and LOQ were est[ab](#page-5-1)lished from the Slope and STEYX by plotting the linearity curve of concentration versus area response for Acetone, Dichloromethane and Cyclohexane. The LOD for Acetone, Dichloromethane and Cyclohexane was found 170, 25 and 118 ppm respectively. The LOQ was found 514, 76 and 359 ppm respectively for Acetone, Dichloromethane and Cyclohexane. The results are presented in Table 6.

Linearity

The linearity of the method was established for Acetone, Dichloromethane a[nd](#page-6-0) Cyclohexane from LOQ to 150% of the target specification concentration by plotting concentration versus peak area response. The method was found linear for Acetone, Dichloromethane and Cyclohexane with correlation coefficient 0.998, 0.998 and 0.999 respectively. The linearity results are tabulated in Table 7. Chromatograms at LOQ and 150 % are shown in Figures 6 and 7. Linearity graphs of Acetone, Dichloromethane and Cyclohexane, are shown in Figures 8, 9 and 10.

Robust[ne](#page-3-5)ss

The robustness of the method was determined by deliber[ate](#page-6-1)[ly](#page-6-2) cha[ngi](#page-7-0)ng the initial oven temperature and nitrogen gas flow rate. Evaluated the system

suitability results at each robustness condition and were found within the acceptance criteria.

Solution stability

Solution stability of the standard solution and test sample solution was established and found to be stable for 48 hours on bench-top (controlled room temperature).

DISCUSSION

As per ICH Q3C, it is mandatory to estimate and control residual solvents used for synthesis, crystallisation and purification of drug substances or API. Several trials were taken on HS-GC to optimise the column dimensions, carrier gas flow, oven temperature, detector temperature, gradient programme, split ratio and standard & test concentrations to achieve good peak shape and better retention and resolution of Acetone, Dichloromethane and Cyclohexane peaks. The developed method was very sensitive and straightforward with a shorter run time. The developed method was validated as per the current method validation guidelines and found suitable.

CONCLUSIONS

A method was developed for the simultaneous estimation of Acetone, Dichloromethane and Cyclohexane in Nitazoxanide API. The developed method was validated as per ICH Q2 and USP <1225> guidelines for system suitability, specificity, precision, accuracy, LOD & LOQ, linearity and robustness. The method validation results were found meeting the acceptance criteria for all parameters. The proposed method is simple, sensitive, selective, accurate and robust for quantitative estimation of Acetone, Dichloromethane and Cyclohexane in Nitazoxanide API by HS-GC and can be used for routine analysis in quality control and research laboratory.

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Conϐlict of interest

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REFERENCES

- Gandhi, S. V., Jadhav, V. Y., Kadukar, S. S. 2008. Simultaneous determination of nitazoxanide and ofloxacin in tablet formulation by ratio spectra derivative spectroscopy. *J. Pharma. Res*, 7:112– 114.
- Harold, M., Nair, M., James, M. 1997. Basic Gas Chromatography. *JOHN WILEY & SONS*.
- ICH 2005. International conference on Hormonisation, Q2(R1) - Validation of analytical procedures: text and methodology, International conference on Hormonisation, IFPMA, Geneva.
- ICH 2016. International conference on Hormonisation, Q3C(R6) - Impurities: Guidelines for Residual Solvents, International conference on Hormonisation, IFPMA, Geneva.
- Korba, B. E., Montero, A. B., Farrar, K., Gaye, K., Mukerjee, S., Ayers, M. S., Rossignol, J.-F. 2008. Nitazoxanide, tizoxanide and other thiazolides are potent inhibitors of hepatitis B virus and hepatitis C virus replication. *Antiviral Research*, 77(1):56–63.
- Kumar, R. S., Nallasivan, P. K., Saravanakumar, S., Kandasamy, C. S., Venkatnarayanan, R. 2009. Simultaneous RP-HPLC estimation of nitazoxanide and oϐloxacin in tablet dosage forms. *Asian Journal of Research in Chemistry*, 2(1):43–45.
- Malesuik, M. D., Goncalves, H. M. L., Paim, C. S., Schapoval, E. E. S., Steppe, M. 2009. LC: Analysis of Photodegradation Kinetics of Nitazoxanide in Pharmaceutical Formulations. *Journal of Chromatographic Science*, 47(9):745–748.
- Mégraud, F., Occhialini, A., Rossignol, J. F. 1998. Nitazoxanide, a Potential Drug for Eradication ofHelicobacter pylori with No Cross-Resistance to Metronidazole. *Antimicrobial Agents and Chemotherapy*, 42(11):2836–2840.
- Narayan, H. J., Mahendra, A. A. 2007. Highperformance liquid chromatography reverses phase method for determination of Nitazomanide from Pharmaceutical Formulations (Oral Suspension and Tablets). *Research Journal of Chemistry and Environment*, 11(3):42–46.
- Narayana, K. L., Manohara, Y. N. 2007. Visible spectrophotometric determination of nitazoxanide in bulk and pharmaceutical dosage forms. *Asian Journal of Chemistry*, 19(4):2527–2530.
- Pandey, S. 2009. Spectrophotometric analysis of nitazoxanide in a single and combined dosage form. *Asian Journal of Chemistry*, 21(6):4149– 4154.
- Rossignol, J.-F., Ayoub, A., Ayers, M. S. 2001. Treatment of Diarrhea Caused byGiardia intesti-

nalisandEntamoeba histolyticaorE. dispar:A Randomized, Double-Blind, Placebo-Controlled Study of Nitazoxanide. *The Journal of Infectious Diseases*, 184(3):381–384.

- Rossignol, J. F., Cavier, R. 1976. A new derivative of 2 benzamido-5-nitrothiazoles. *Chem. Abstr*, 83:28– 28.
- Sitaramaraju, Y., Riadi, A., D'Autry, W., Wolfs, K., Hoogmartens, J., Schepdael, A. V., Adams, E. 2008. Evaluation of the European Pharmacopoeia method for control of residual solvents in some antibiotics. *Journal of Pharmaceutical and Biomedical Analysis*, 48(1):113–119.
- USP 2018a. The United States Pharmacopoeia 41, The National Formulary 36, General Chapter <1225> - Validation of Compendial Procedures, United States Pharmacopeia Convention.
- USP 2018b. The United States Pharmacopoeia 41, The National Formulary 36, General Chapter <621> - Chromatography, United States Pharmacopeia Convention.